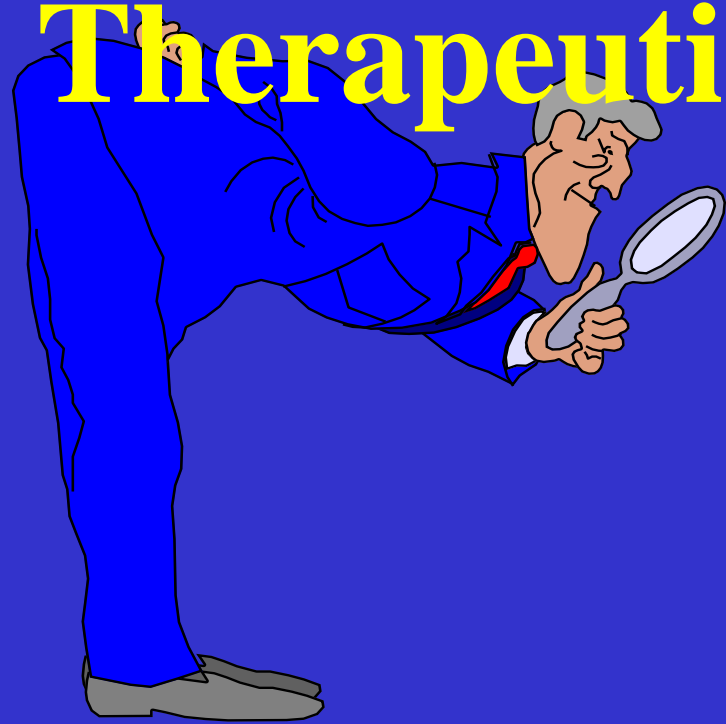


Clinical Pharmacokinetics and

Therapeutic Drug Monitoring



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Outline

- Review of Concepts
 - Dose, Dosage, Clearance, Half-Life, Volume of Distribution
 - loading dose, maintenance dose and time to achieve steady state plasma concentration (C_{pss})
- Therapeutic drug Monitoring
- Cases

- Discussion/Questions

What is clinical pharmacokinetics ?

- Study of the time course of a drug's movement through the body.
- Understanding of what the body does to (or with) the drug.
- Application of Therapeutic Drug Monitoring (TDM) and individualization of drug therapy.

WHY BE CONCERNED ABOUT CLINICAL PHARMACOKINETICS AND DOSAGE REGIMENS?

Pharmacokinetics and Dosage Regimens
Determine:

- How much drug is in the body at any given time
 - How long it takes to reach a constant level of drug in the body during chronic drug administration
 - How long it takes for the body to rid itself of drug once intake of drug has stopped

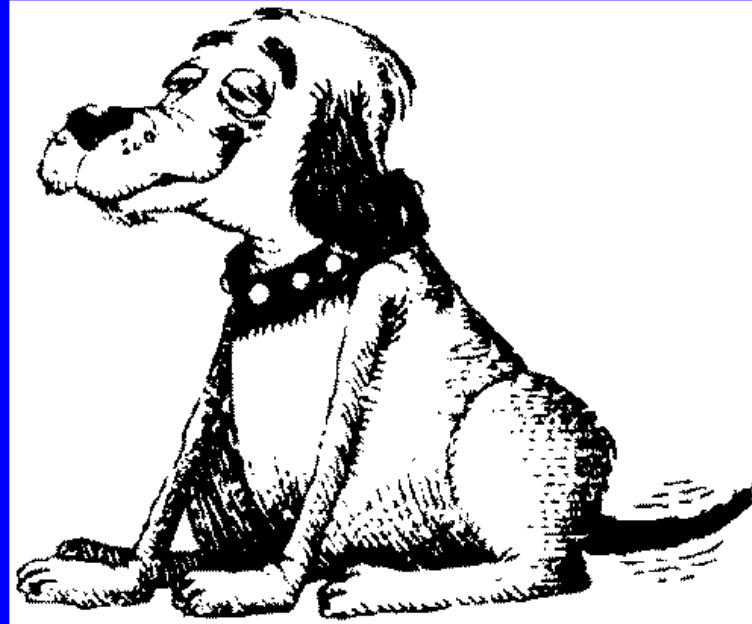
Basic Parameters

- In the next few slides the basic concepts and parameters will be described and explained.
- In pharmacokinetics the body is represented as a single or multiple compartments in to which the drug is distributed.
- Some of the parameters are therefore a little abstract as we know the body is much more complicated!

WARNING!!
THE STUDY OF PHARMACOKINETICS
MAKES SOME PEOPLE ANXIOUS



BUT RELAX,



Pharmacokinetics Can Be Made FRIENDLY

MAJOR CONCEPT #1
CONCEPT OF VOLUME OF
DISTRIBUTION
(V_D) OF DRUGS

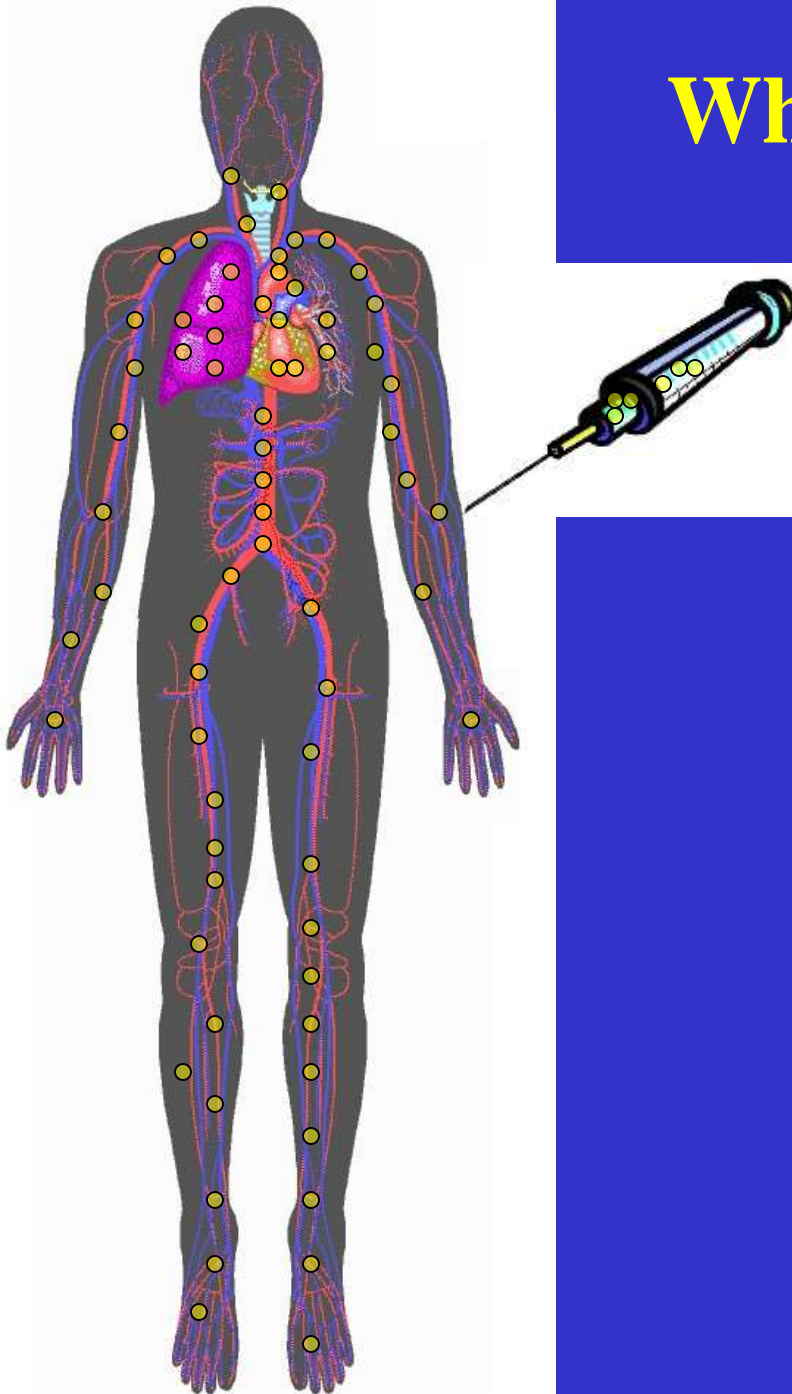
Opening Question:

Mr. JR receives 500 mg of Levofloxacin by intravenous bolus (over 1 minute) and the serum concentration is measured immediately and found to be 5.0 mg/L.

What is the apparent volume of distribution?

What percent of drug is located in serum?

Where does the drug go?



Is it confined to blood

or

is it mostly in the blood

or

is it largely confined to tissues?

Barriers to Distribution

1. GI Tract

⇒ Intestinal wall prevents absorption
... not all drugs are absorbed

2. Vascular walls

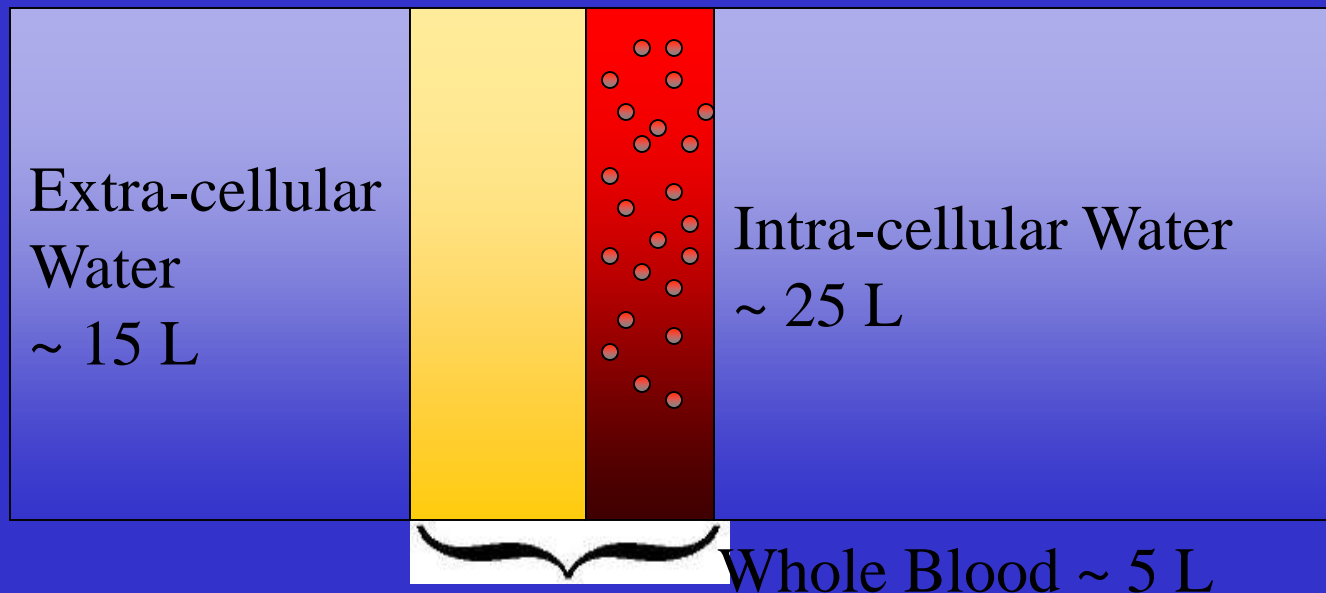
⇒ limits “escape” from serum / blood

3. Cellular walls

⇒ limits “free” movement within the body

Plasma or Serum ~ 55% ↓

↓ RBC's, ~ 45% of whole blood

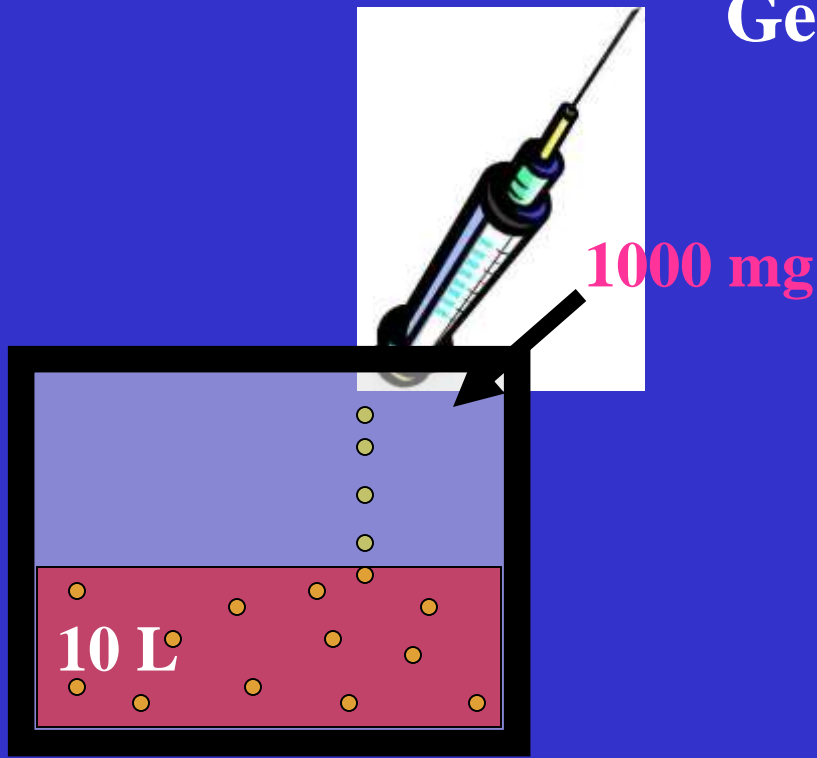


Body Water¹

Tissue	% Water	% Weight	Water per 70 kg (L)
Skin	72	18	9.1
Muscle	75	42	22.1
Brain	75	2	1.1
Skelton	22	16	2.5
Adipose	10	~10	0.7
Other		12	6.5
Total		100	42

1. Skelton, H. Arch Int. Med 1927; 40: 140.

General Principles of Distribution



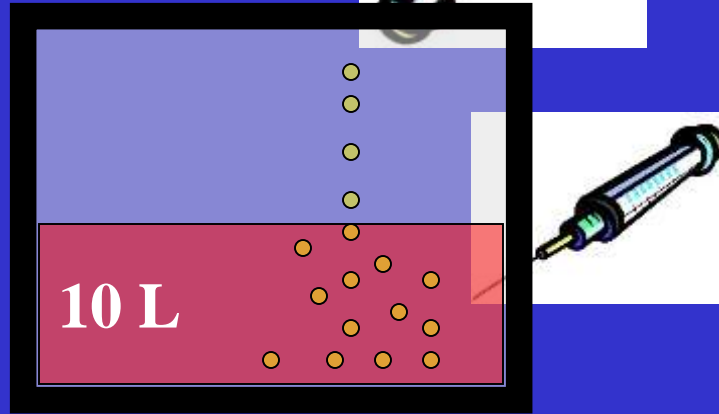
If you add 1000 mg of a drug to 10 L of water, what is the final concentration?

Following complete mixing
Concentration (C) = 1000 mg / 10L
= 100 mg/L

$$C = \text{Amount} / \text{volume}$$

$$\text{Volume} = \text{Amount} / C$$

General Principles of Distribution



1000 mg

If you add 1000 mg of a drug to 10 L of water, what is the final concentration?

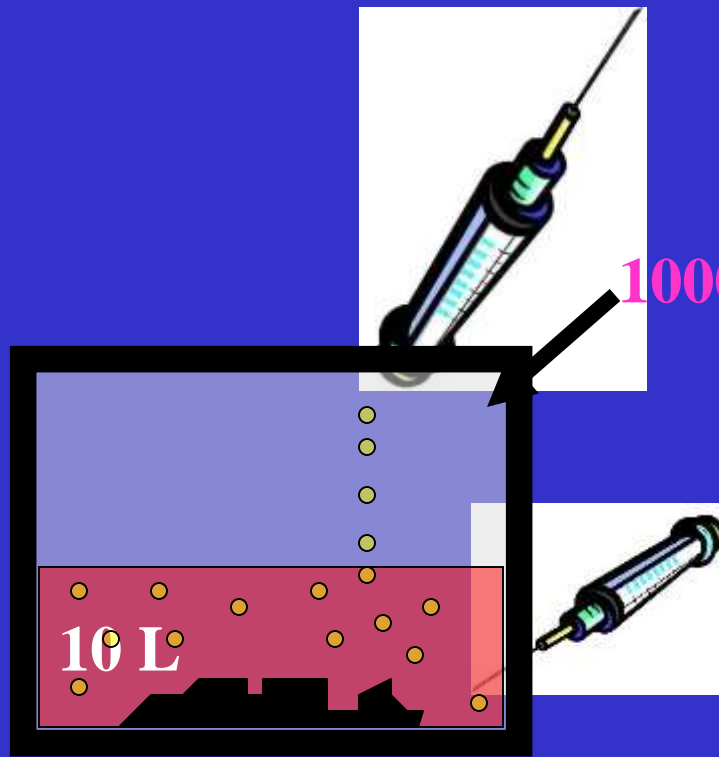
$$C = \text{Amount} / \text{volume}$$

$$\text{Volume} = \text{Amount} / C$$

Could the concentration change as a function of time after addition?

If you knew that you had added 1000 mg of drug and then drew a sample from a corner of the vessel before complete mixing occurred, what would you conclude?

General Principles of Distribution



Charcoal

Again you add 1000 mg of a drug to 10 L of water, but now there is some charcoal in the water that may bind the drug.

The observed concentration after complete mixing is 50 mg/L

Since

Volume = Amount / C

then the

apparent volume of distribution is:

= 1000 mg/50 mg/L

= 20L ...???

BUT The real volume is 10L

General Principles of Distribution

Again you add 1000 mg of a drug but this time in addition to the charcoal and 10L of water there is 1 L of oil.

You measure the concentration in the oil (150 mg/L) and in the water (25 mg/L).

Now calculate the volume:

Based on the concentration in the water

$$\text{Volume} = \text{Amount} / C$$

then the

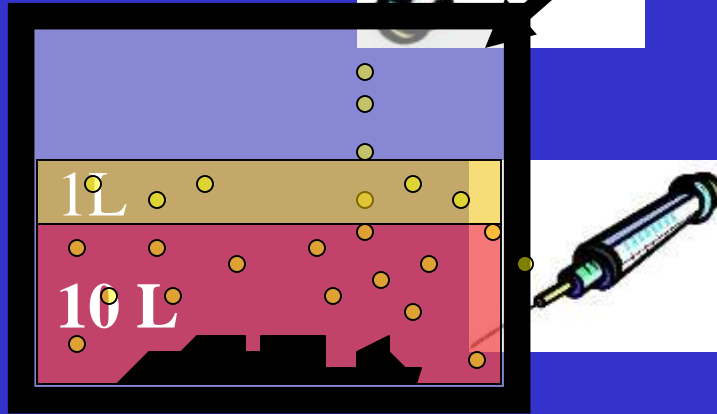
apparent volume of distribution is:

$$= 1000 \text{ mg} / 25 \text{ mg/L}$$

$$= 40\text{L}$$

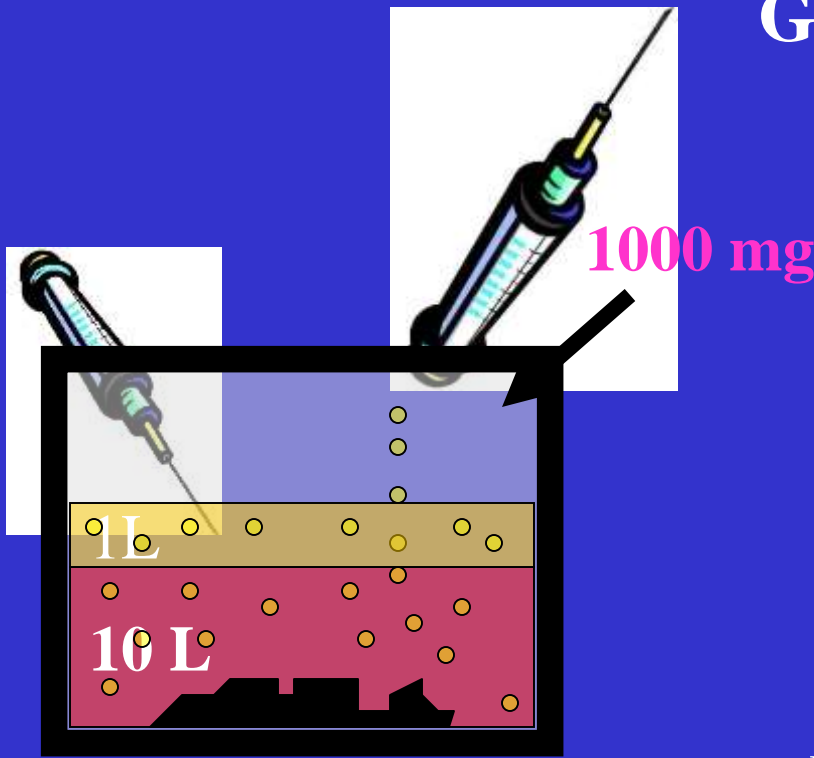


1000 mg



Charcoal

General Principles of Distribution



Again you add 1000 mg of a drug but this time in addition to the charcoal and 10L of water there is 1 L of oil.

You measure the concentration in the oil (150 mg/L) and in the water (25 mg/L).

Now calculate the volume:

Based on the concentration in the oil

$$\text{Volume} = \text{Amount} / C$$

then the

apparent volume of distribution is:

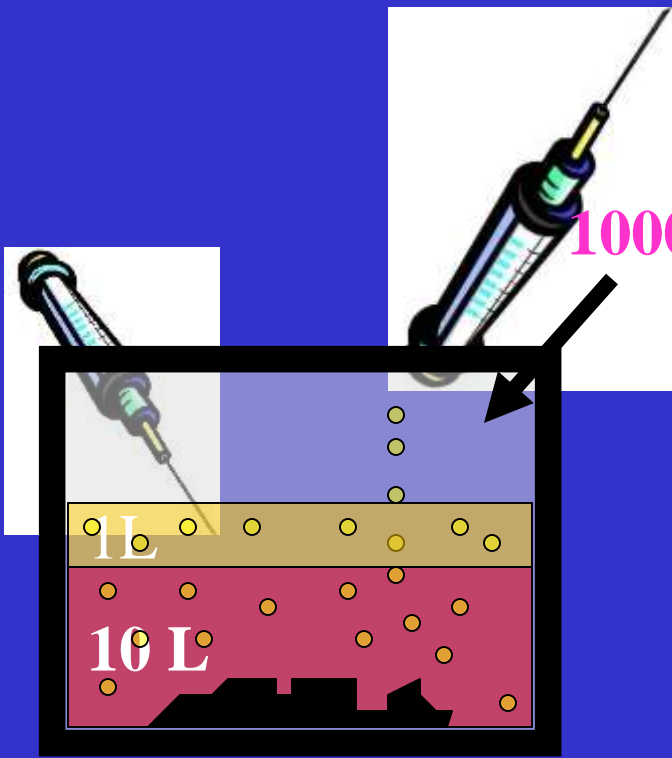
$$= 1000 \text{ mg} / 150 \text{ mg/L}$$

$$= 6.66 \text{ L}$$

Charcoal

General Principles of Distribution

Mass Balance



Water:

Concentration 25 mg/L

True Volume: 10L

Amount of Drug = 250 mg

Apparent Volume = 40L

Oil:

Concentration 150 mg/L

True Volume: 1L

Amount of Drug = 150 mg

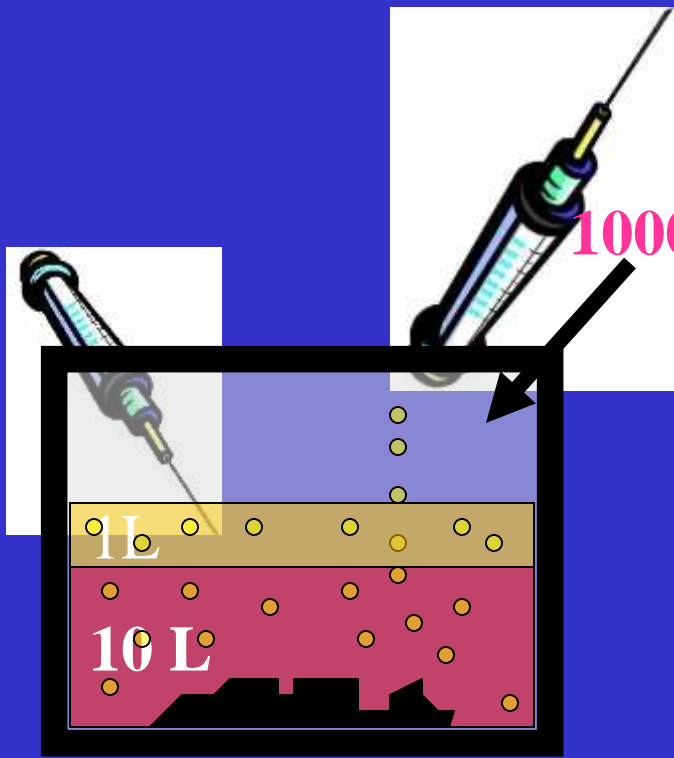
Apparent Volume = 6.66 L

Charcoal

Charcoal: (therefore)

Amount = 600 mg

General Principles of Distribution



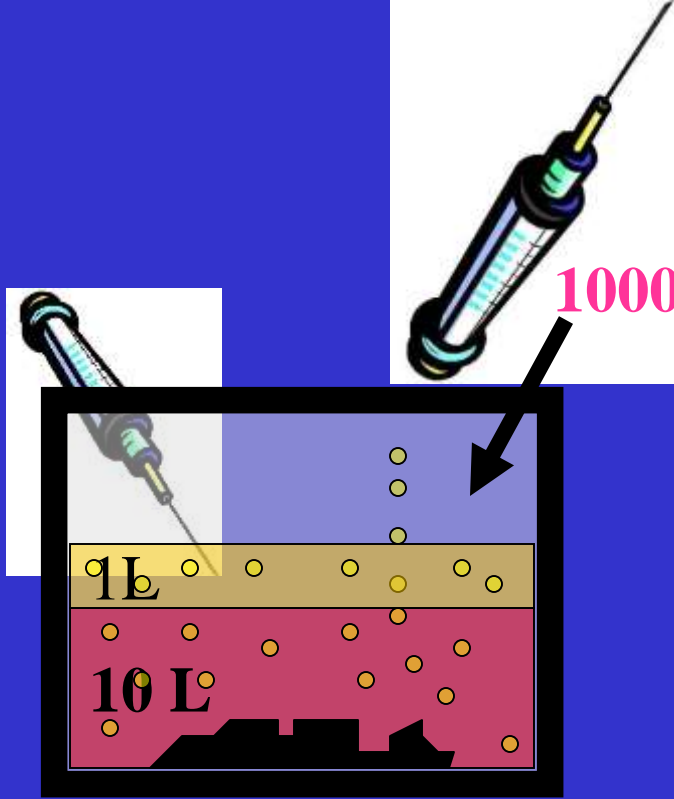
1000 mg

Conclusions

1. The calculated Apparent Volume depends on the fluid being sampled.
2. The volume depends on the host, and the physical/chemical properties of the drug or metabolite
3. The calculated Apparent Volume rarely reflects a real physiologic volume.

Charcoal

General Principles of Distribution



Charcoal

...so what is the Apparent Volume of Distribution?

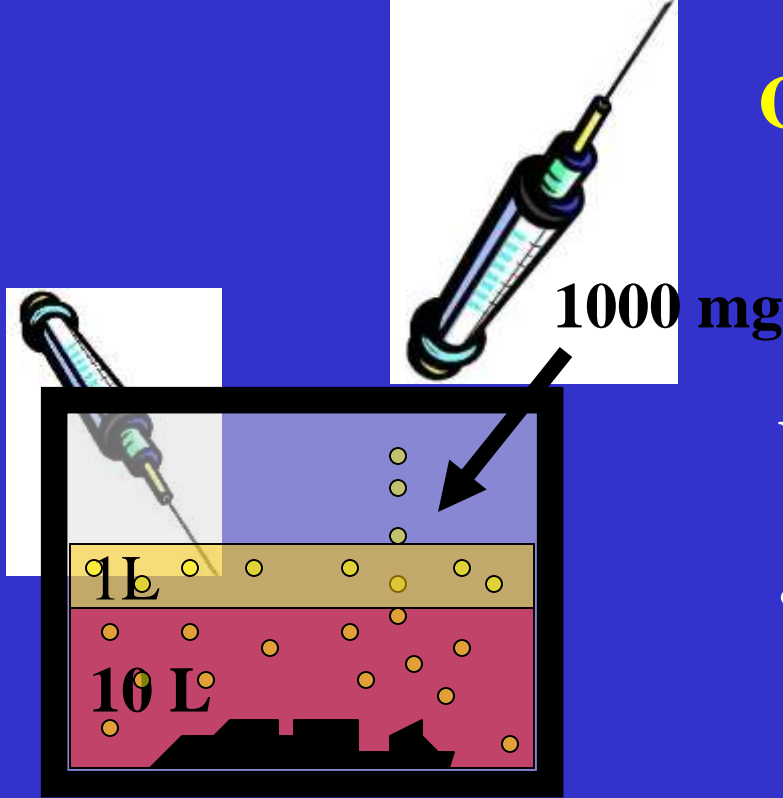
it is the volume of *sampled* fluid need to *account* for the total amount of drug in the body ... at distribution equilibrium ... (following complete mixing).

The volume is not associated with a particular space or anatomical area or tissue.

It is a proportionality constant relating concentration and amount in the body.

General Principles of Distribution

...so, if it is not real, how useful is it?



Uses:

- It tells us how much drug must be added to the body so as to achieve a specified concentration in the sampled fluid.
- In a general way it tells us where the drug is stored in the body or where it might be found.

Charcoal

L/70 kg

50,000

20,000

10,000

5,000

1,000

500

100

50

10

5

Quinacrine

Chloroquine

Nortriptyline
Digoxin

Propranolol
Quinidine

Quinolones (1- 2 L/kg), Tetracycline

Phenobarbital

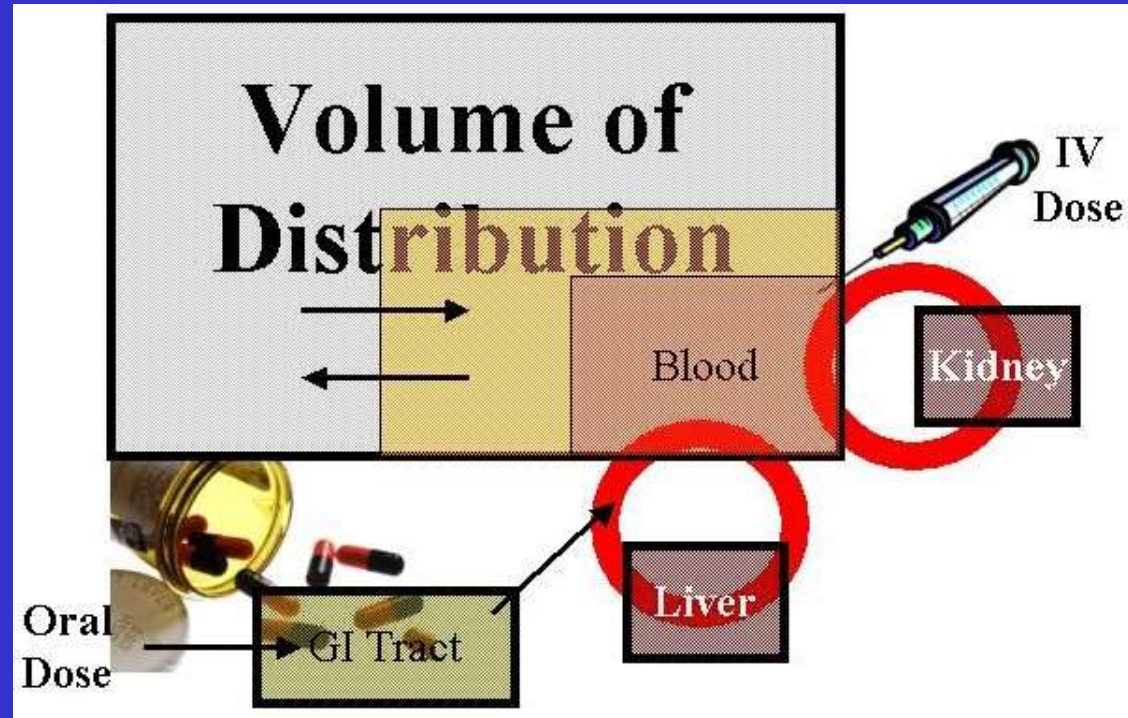
Phenytoin

Theophylline (0.45 L/kg)

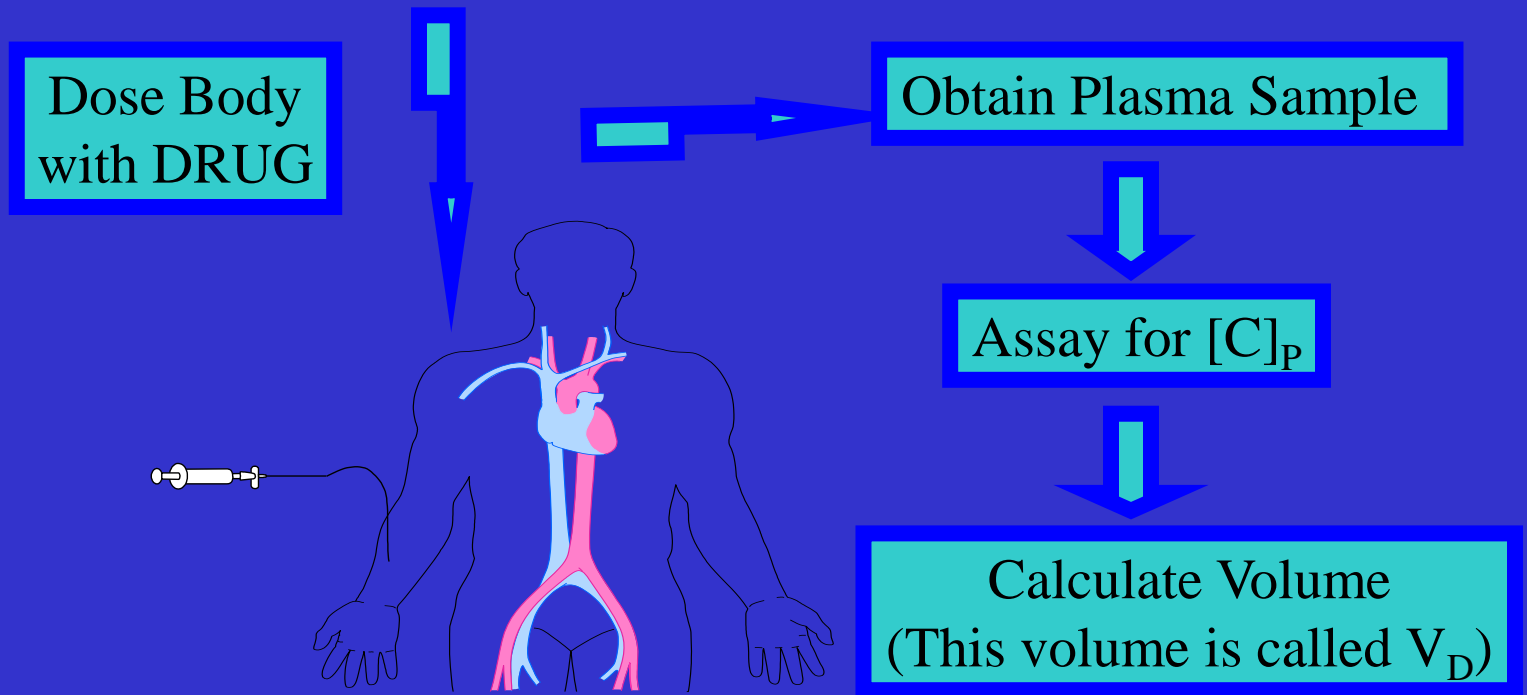
Aminoglycosides (0.25 L/kg)

ASA

Warfarin



CONCEPT OF VOLUME OF DISTRIBUTION OF DRUGS: DEFINITION OF V_D



By DEFINITION: $V_D = D/[C]_P$

(where D is amount of drug in body and $[C]_P$ is concentration of drug in plasma)

Equations

$$\text{Conc} = \text{Dose} / V$$

$$V = \text{Dose}/\text{Conc}$$

Answers.

1. What is the Volume of distribution of levo?

$$\text{Dose} =$$

$$\text{Initial []} =$$

$$\text{Levo Volume:} =$$

2. Proportion in Serum: =

Equations

$$\text{Conc} = \text{Dose} / V$$

$$V = \text{Dose} / \text{Conc}$$

Answers.

1. What is the Volume of distribution of levo?

$$\text{Dose} = 500 \text{ mg}$$

$$\text{Initial []} = 5 \text{ } \mu\text{g/mL (mg/L)}$$

$$\begin{aligned} \text{Levo Volume:} &= \text{Dose} / \text{Conc} \\ &= 500 / 5 \\ &= 100 \text{ L} \end{aligned}$$

2. Proportion in Serum: =

Equations

$$\text{Conc} = \text{Dose} / V$$

$$V = \text{Dose}/\text{Conc}$$

Answers.

2. Proportion in Serum:

Weight : 80 kg

Levo Volume: 500 mg/ 5 mg/L

Blood Volume: 8% of body weight

Hematocrit: 0.45

Blood volume: =

Serum (55%): =

Levo Volume: =

Proportion in Serum: =

Equations

$$\text{Conc} = \text{Dose} / V$$

$$V = \text{Dose}/\text{Conc}$$

Answers.

2. Proportion in Serum:

Weight : 80 kg

Levo Volume: 500 mg/ 5 mg/L

Blood Volume: 8% of body weight

Hematocrit: 0.45

Blood volume: = 0.08 x 80 kg = 6.4 L

Serum (55%): = 6.4 x 0.55 = 3.5 L

Levo Volume: = 500 mg/ 5 mg/L = 100 L

Proportion in Serum: =

Equations

$$\text{Conc} = \text{Dose} / V$$

$$V = \text{Dose}/\text{Conc}$$

Answers.

2. Proportion in Serum:

Weight : 80 kg

Levo Volume: 500 mg/ 5 mg/L

Blood Volume: 8% of body weight

Hematocrit: 0.45

Blood volume: = 0.8 x 80 kg = 6.4 L

Serum (55%): = 6.4 x 0.55 = 3.5 L

Levo Volume: = 500 mg/ 5 mg/L = 100 L

Proportion in Serum: = 3.5 L/100.0 L
= 3.5%

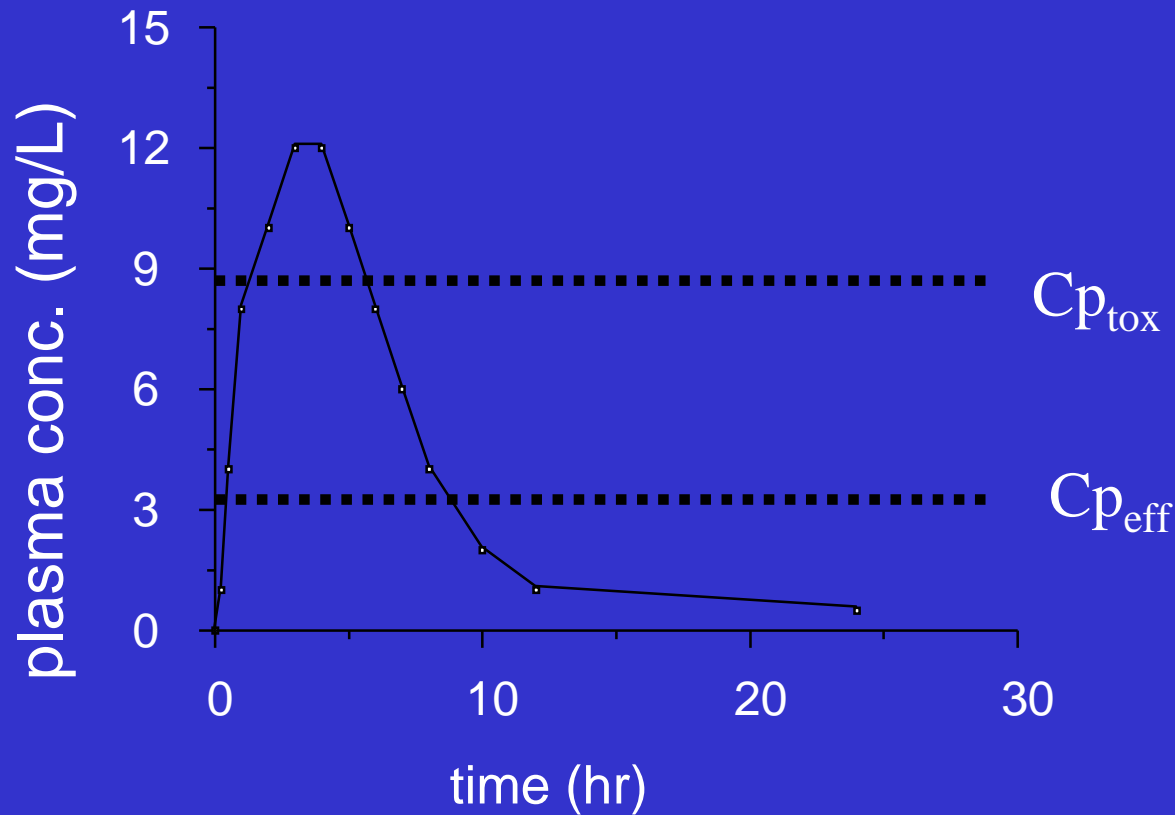
- how to make sense of the previous lesson

or

- why do I have to learn this stuff?

drug dosing calculations

Concepts: *target plasma concentration*
therapeutic window



- Therapeutic window: Difference between the minimum effective concentrations (MEC) required for a desired response and one that produces an adverse effect.
- For some drugs it is small (only two- to three-fold difference)
- E.g. digoxin, theophylline, lidocaine, aminoglycosides, cyclosporine, anticonvulsants.

Loading Dose (DL) = a *dose* of drug sufficient to produce a plasma concentration of drug that would fall within the therapeutic window after only one or very few doses over a very short interval.

Information needed to calculate a DL

- Volume of distribution, usually in L or L/Kg - from literature -
- Desired concentration - from literature

CONCEPT OF VOLUME OF DISTRIBUTION OF DRUGS: INTRODUCTION TO V_D

By DEFINITION: $V_D = D/[C]_P$

Rearranging: $D = V_D \times [C]_P$

Suppose you want a certain desirable $[C]_P$, call it $[C]_{P(\text{target})}$

Substituting $[C]_{P(\text{target})}$ for $[C]_P$: $D_{\text{target}} = V_D \times [C]_{P(\text{target})}$

Where D_{target} is the amount of drug in body required to achieve a given $[C]_{P(\text{target})}$

CONCEPT OF VOLUME OF DISTRIBUTION OF DRUGS: INTRODUCTION TO V_D

If patient has no drug in body to begin with, then can administer an amount (called “Loading Dose”) to achieve a given D_{target} and $[C]_{P(\text{target})}$

For the *oral* loading dose we have to take the *fraction bioavailable* into account (0-1)

$$L = C_p \cdot V_d / F$$

Loading dose

- $LD = \text{target } C_p \cdot V_d / F$

e.g. lidocaine $t_{1/2} = 1-2 \text{ h}$

Post MI arrhythmias – life threatening – can't wait 4-8 h –
LD is used

- Another example: Digoxin for heart failure
- If only maintenance dose is given it takes 10 days $t_{1/2} = 61 \text{ h}$

$$DL = 1.5 \text{ ng/ml} \times 580 / 0.7 = 1243 \text{ micg} \sim 1 \text{ mg}$$

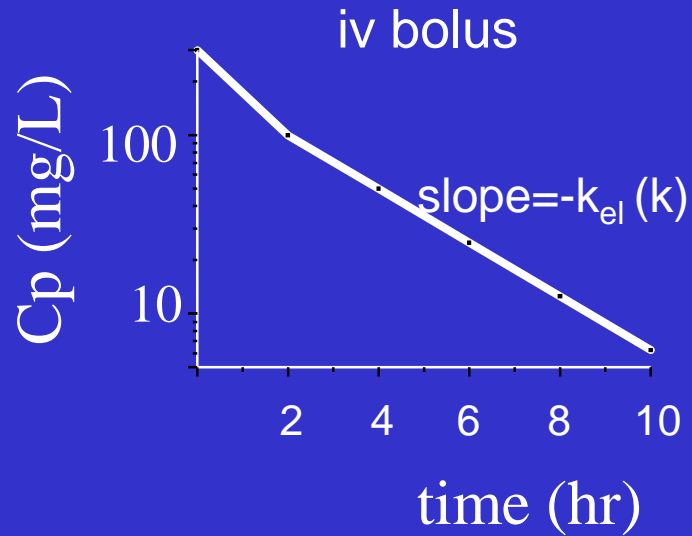
Can be given iv divided – 0.5 mg, aft 6-8 h 0.25 mg, aft 6h
0.125 mg and then 0.125 mg (to avoid over digitalization
and toxicity)

Half-life ($t_{1/2}$)

time interval after which the concentration is half that at the beginning of the time interval

has real-life meaning *only* in first-order kinetics!

half life ($t^{1/2}$)



the *time* taken for the C_p to fall by half

$$t^{1/2} = 0.693/k$$

where k is the rate constant (in hrs^{-1})

most drugs follow first order kinetics:

rate of change of drug in the body = $-k_{el}$ · Amount of Drug in the body

where k_{el} (k) is the *rate constant* of elimination

a few drugs (e.g. ethanol) follow 0 order kinetics:

rate of change = k

a few drugs (e.g. phenytoin) follow 1st order kinetics:

at low conc. and 0 order at high conc.

zero-order

constant *amount* of drug eliminated per unit of time

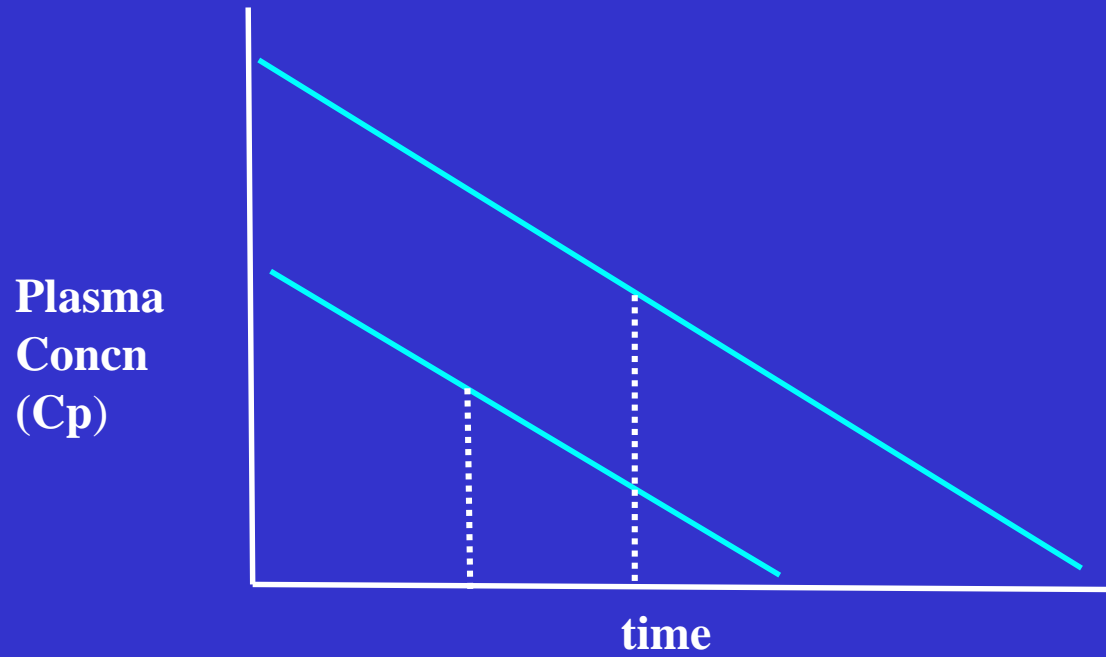
elimination rate constant has units of weight/time:

$K_e = 50 \text{ mg/h}$, for example

typical drug: ethanol

Zero order elimination

Half life varies with concentration



Designing a Dosage Regimen

Capacity-Limited Metabolism
(Also called “Zero Order Kinetics”)

- An infrequent, but important phenomenon
- Clearance is not constant with respect to C_p because metabolizing enzymes are saturated at “therapeutic concentrations”
- Rate of drug elimination is fixed and cannot use clearance to calculate dosage regimen
- For such drugs, daily dose should not exceed fixed rate of elimination

Designing a Dosage Regimen

Ethanol is Eliminated by “Zero Order Kinetics”

- For average adult, rate of metabolism is 10 g/hr
- 45 ml of whiskey contains 14 g of ethanol
- If drink 45 ml of whiskey every hr, will accumulate 4 g ethanol/hr and develop coma in 48 hr
- However, can drink 30 ml whiskey (9 g ethanol) every hr with impunity

first-order

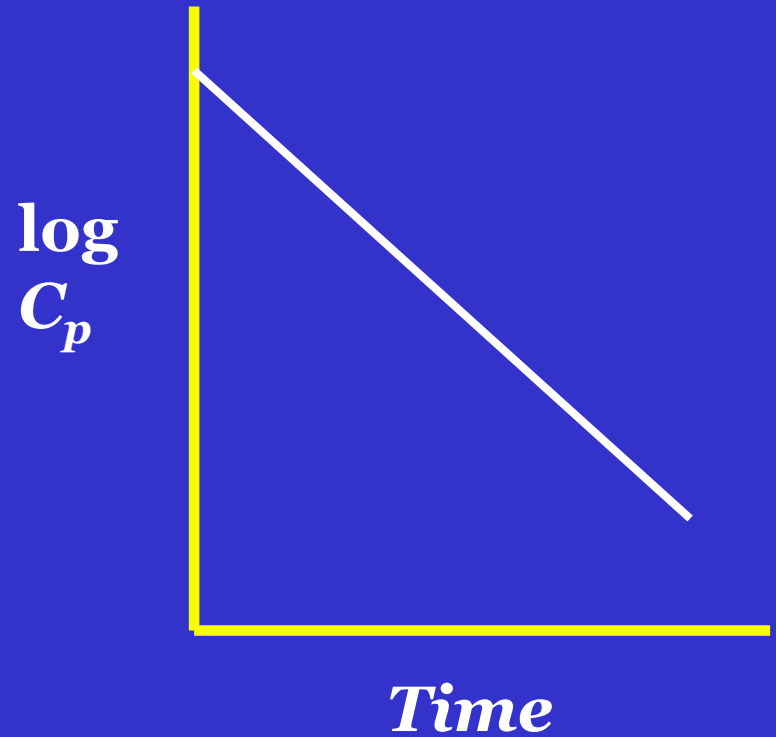
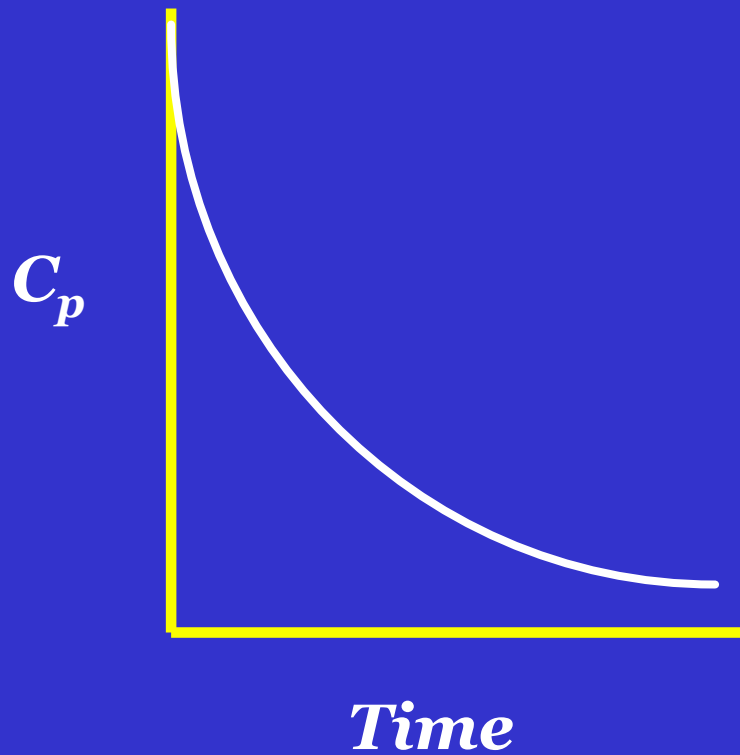
constant *fraction or percent* of drug eliminated per unit of time

elimination rate constant has units of weight/time:

$K_e = 0.25/\text{h}$ or 0.25 h^{-1} , for example.

typical drug: theophylline, most other drugs

first-order elimination following single IV bolus dose



Special properties of first-order kinetics

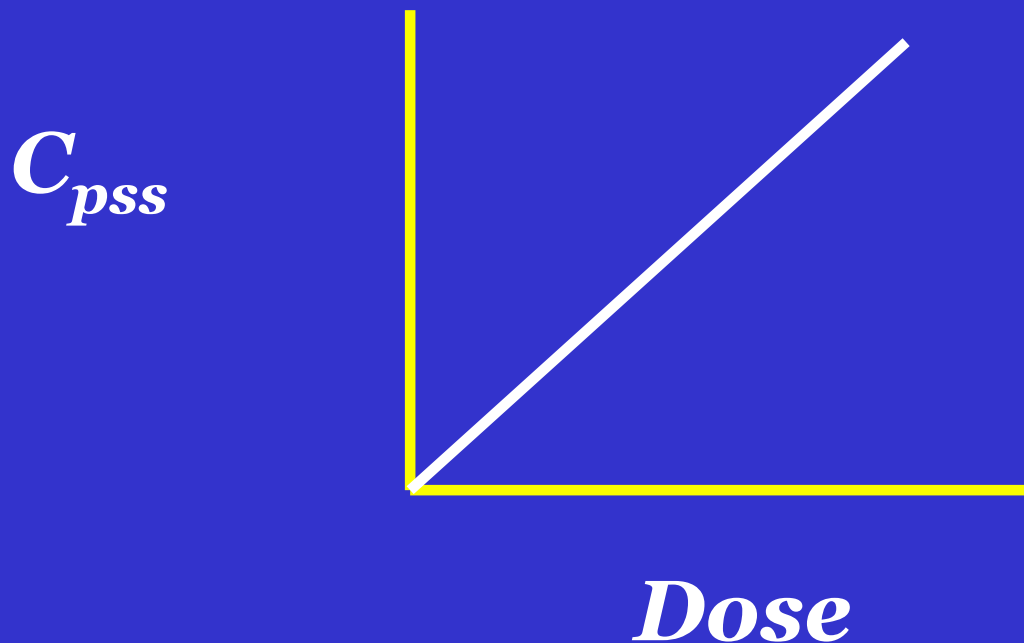
C_p is 90% of the way to new steady-state level after change of dose in $3 t_{1/2}$,
96% in $5 t_{1/2}$

this is true regardless of
initial concentration
final concentration
dose

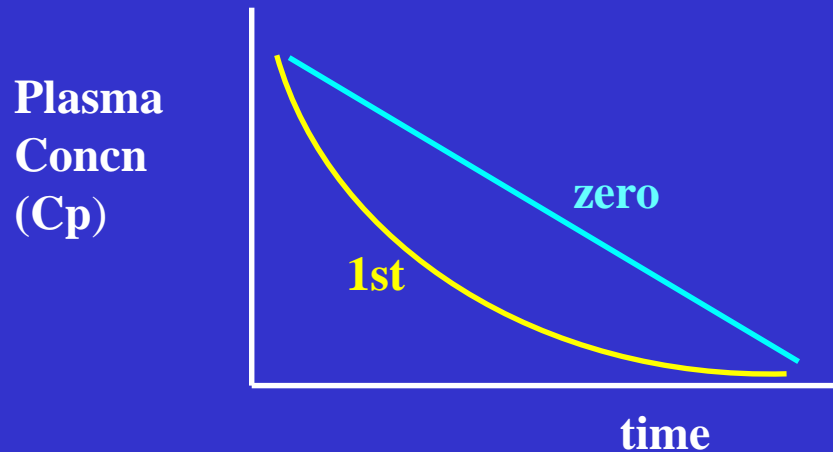
*but **only** for first-order kinetics !*

First-order is also known as *linear kinetics*:

there is a linear relationship between dose and steady-state concentration



Revision of pharmacokinetic terms



1st order elimination

rate of elimination depends on plasma concentration

$$C = C_0 e^{-kt} \quad (k = \text{rate constant of elimination})$$

Half life ($t_{1/2}$)

time for plasma concentration to fall by 50%

Zero order elimination

rate of elimination is constant and independent of plasma concentration

A patient's peak serum phenobarbital level is 12.0 ug/mL at steady state on a dose of 60 mg/day. If we want a level of 20 ug/mL, what new dose should we order if we know that phenobarbital has linear kinetics?

$$\frac{12.0 \text{ ug/mL}}{20 \text{ ug/mL}} = \frac{60 \text{ mg/day}}{X}$$

$$X = 100 \text{ mg/day}$$

If we change the dose to 90 mg/day, when can we get another plasma level to check our calculations?

In adults, the half-life averages 100 hours. In 3 half-lives (300 h, or 12.5 days) we will be 90% of the way from 12.2 ug/mL to the new steady-state level (we hope it's 20 ug/mL). This is close enough to get a level. Reschedule the patient to come back in 2 weeks for this level.

**change
dose**



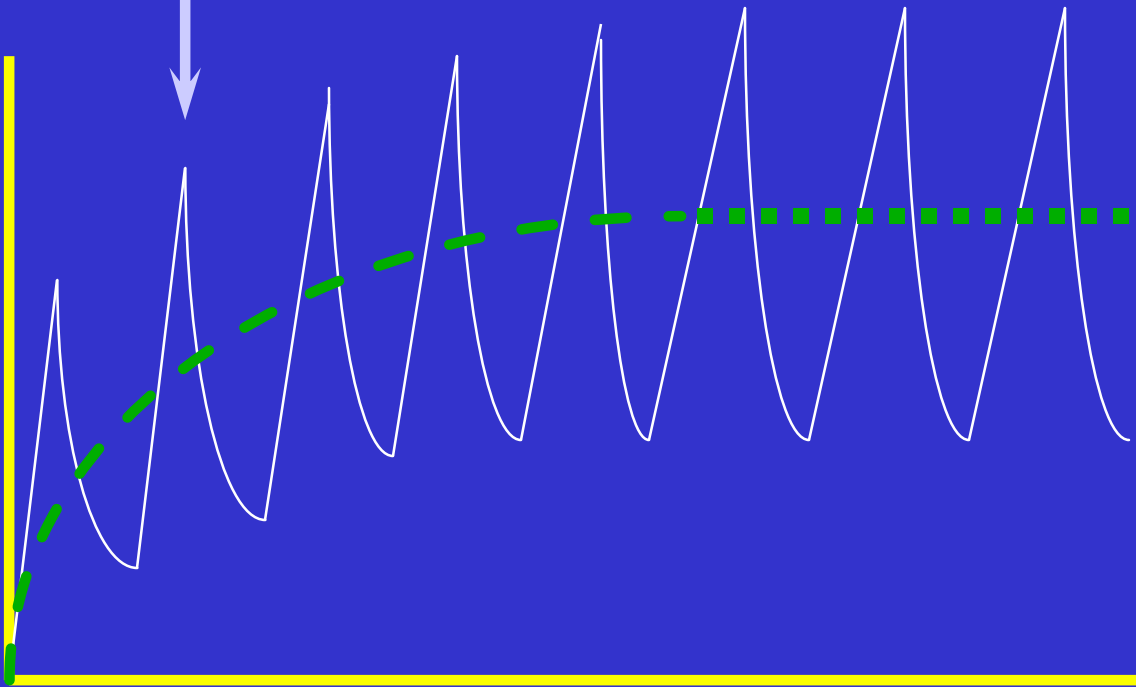
**level
not OK**



level OK



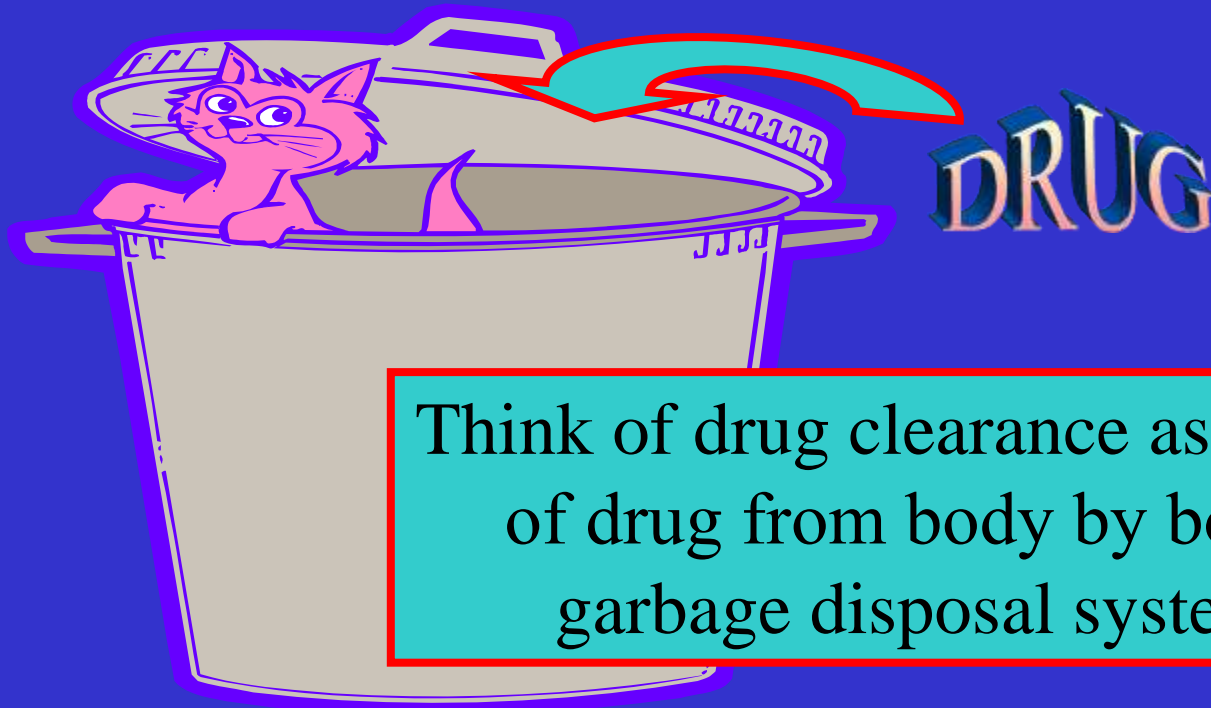
C_p



Time

MAJOR CONCEPT #2

CONCEPT OF DRUG CLEARANCE (Cl)



clearance

defined as the volume of blood from which drug is irreversibly removed per unit time.

ml/min

can calculate: $UV/P = \text{excretion rate/plasma concentration}$

at *steady state*: rate of excretion = rate in (the dose, M)

$$M = \text{rate of excretion} = Cl_{\text{drug}} \cdot C_p$$

Clearance

- Clearance is the **VOLUME** of plasma that can be freed of drug per unit time, i.e., gives estimate of function of organs of elimination and rate of removal of drug from the body

- $$CL = \frac{\text{Rate of Elimination (mg/hr)}}{\text{Concentration (mg/L)}} = \text{vol/time}$$

CONCEPT OF DRUG CLEARANCE (Cl): DEFINITION OF Cl

By Definition:

$$Cl = \frac{\text{Rate of Drug Elimination}}{[C]_P}$$

Units of Cl:

$$\frac{\text{Amount/Time}}{\text{Amount/Volume}} = \frac{\text{Volume}}{\text{Time}}$$

CONCEPT OF DRUG CLEARANCE: DEFINITION OF Cl

Example:

Rate of Drug Elimination = 10 mg/hr

$$[D]_p = 4 \text{ mg/L}$$

$$Cl = \frac{10 \text{ mg/hr}}{4 \text{ mg/L}} = 2.5 \text{ L/hr}$$

the maintenance dose

- **Maintenance Dose (DM)** = The dose needed to maintain the concentration within the therapeutic window when given repeatedly at a constant interval

$$M = Cl_{\text{drug}} \cdot C_p$$

for oral dosing

$$M = Cl_{\text{drug}} \cdot C_p / F$$

Why clearance and V_d ?

- clearance and volume of distribution are independent variables.
- they determine the half-life ($Cl = k * V_d$)

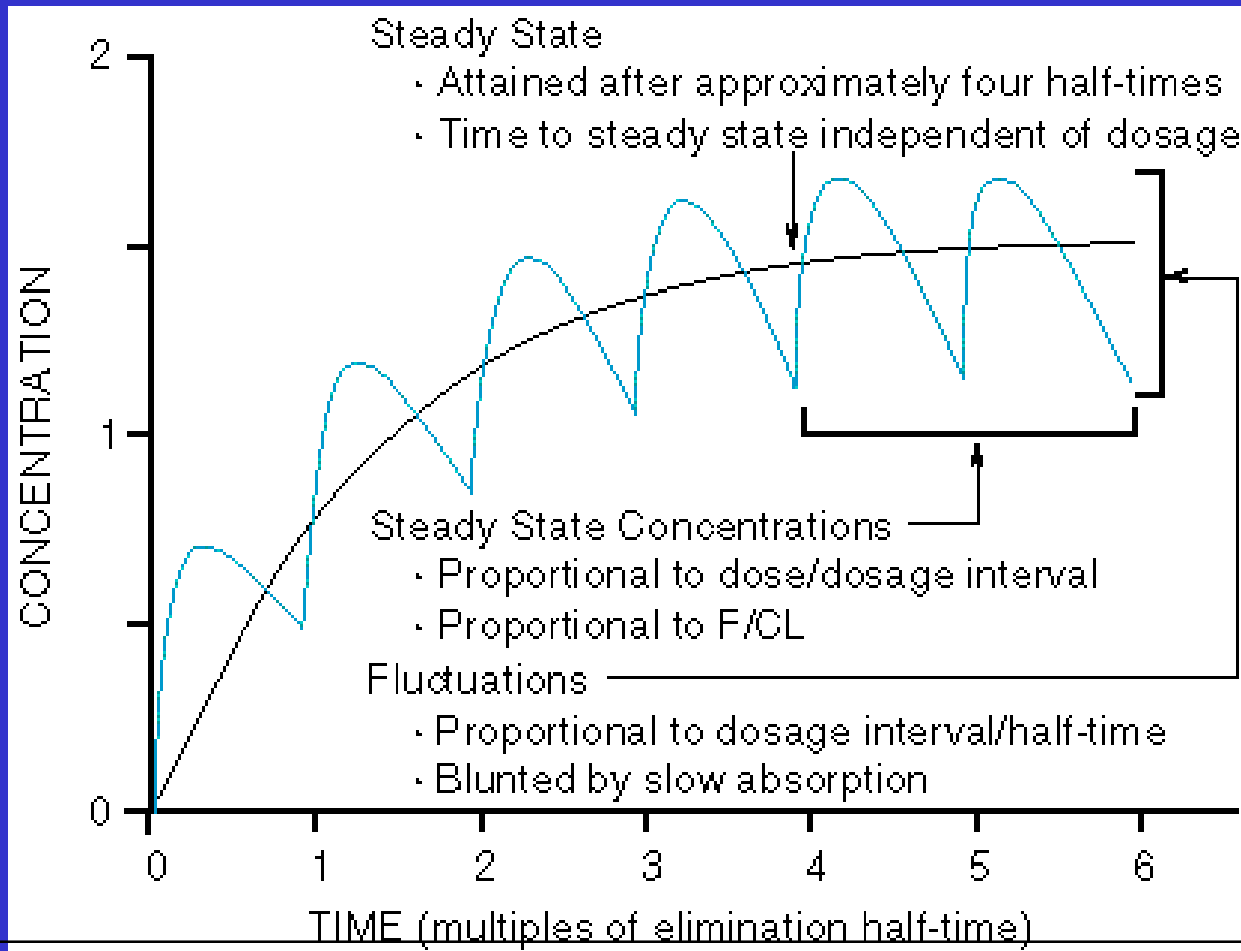
$$t_{1/2} = \frac{0.693}{k}$$

or

$$k = \frac{0.693}{t_{1/2}}$$

$$t_{1/2} = \frac{0.693 V_d}{Cl}$$

oral dosing



the time taken
to reach
steady state
depends only
on $t^{1/2}$

(= 4-5 x $t^{1/2}$)

$$C_{pss} = \frac{F \cdot \text{dose}}{CL \cdot T}$$

$$M (\text{average } C_p) = \frac{Cl \cdot C_p}{F}$$

for drugs with short $t^{1/2}$, must dose
frequently

Time to steady-state or elimination is independent of dosage

Time required to reach steady state

Depends on elimination rate

Requires 5 elimination half-lives to reach 97% of steady state

Requires 5 elimination half-lives to eliminate 97% of drug

number of half-lives	Drug	remaining
0	100	500
1	50	250
2	25	125
3	12.5	62.5
4	6.25	31.25
5	3.125	15.625

summary

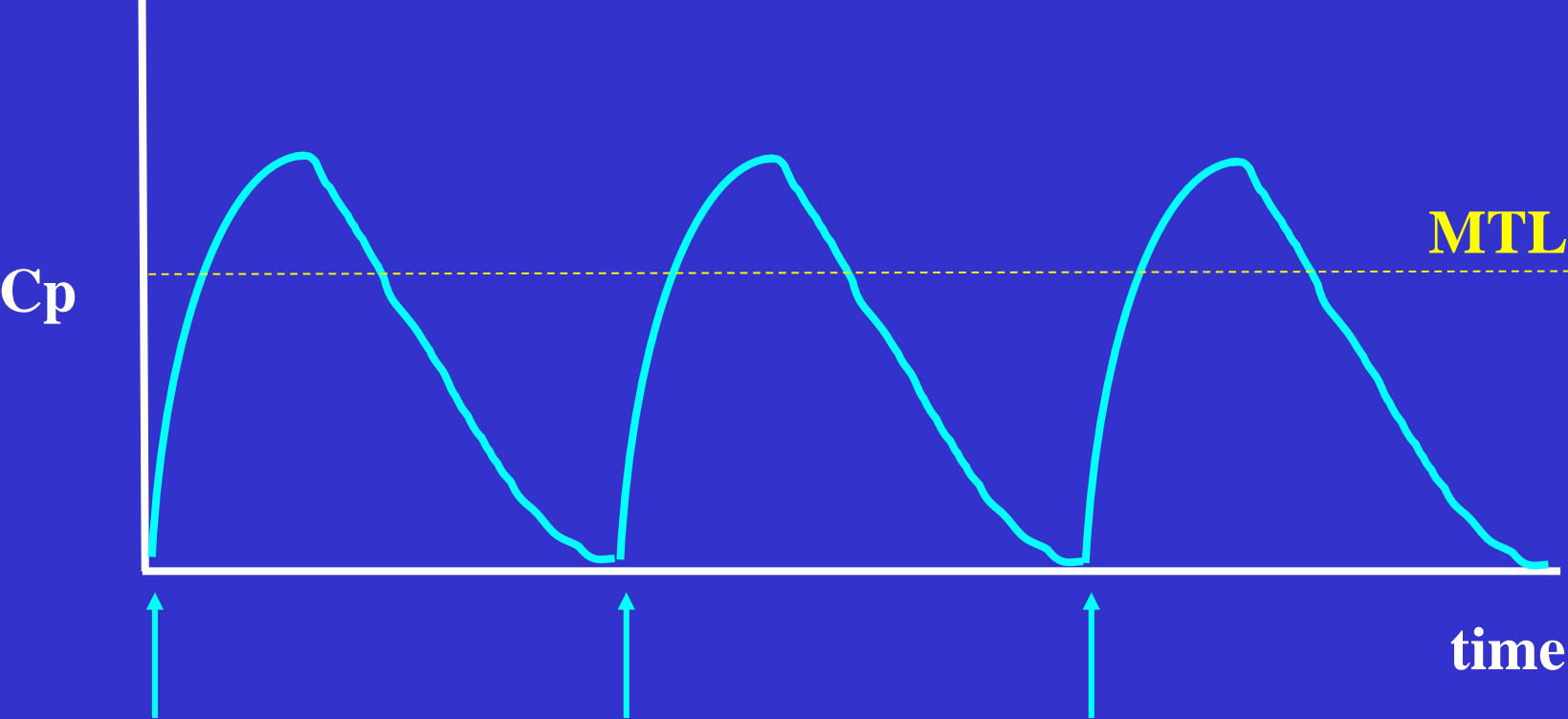
- drug dosing should aim for the *target plasma concentration*.
- the *volume of distribution* is useful in calculating the *loading dose*
- the *clearance* is useful in calculating the *maintenance dose*
- the time to reach steady state depends only on the *half life*

Drug dosing

Important factors

- concentration of drug in plasma
- rate of drug elimination
- rate of drug absorption

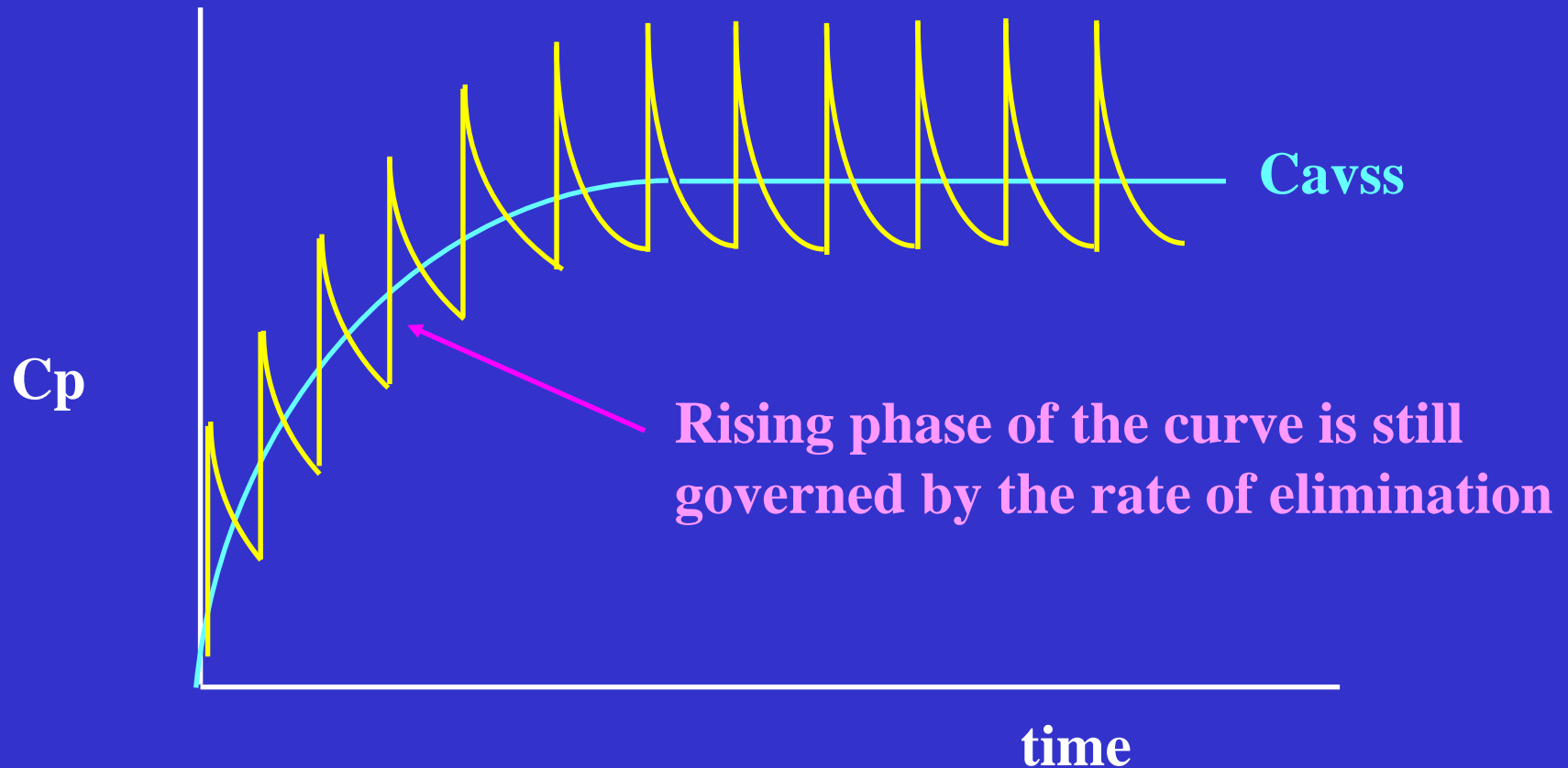
Dosing interval



Multiple dosing

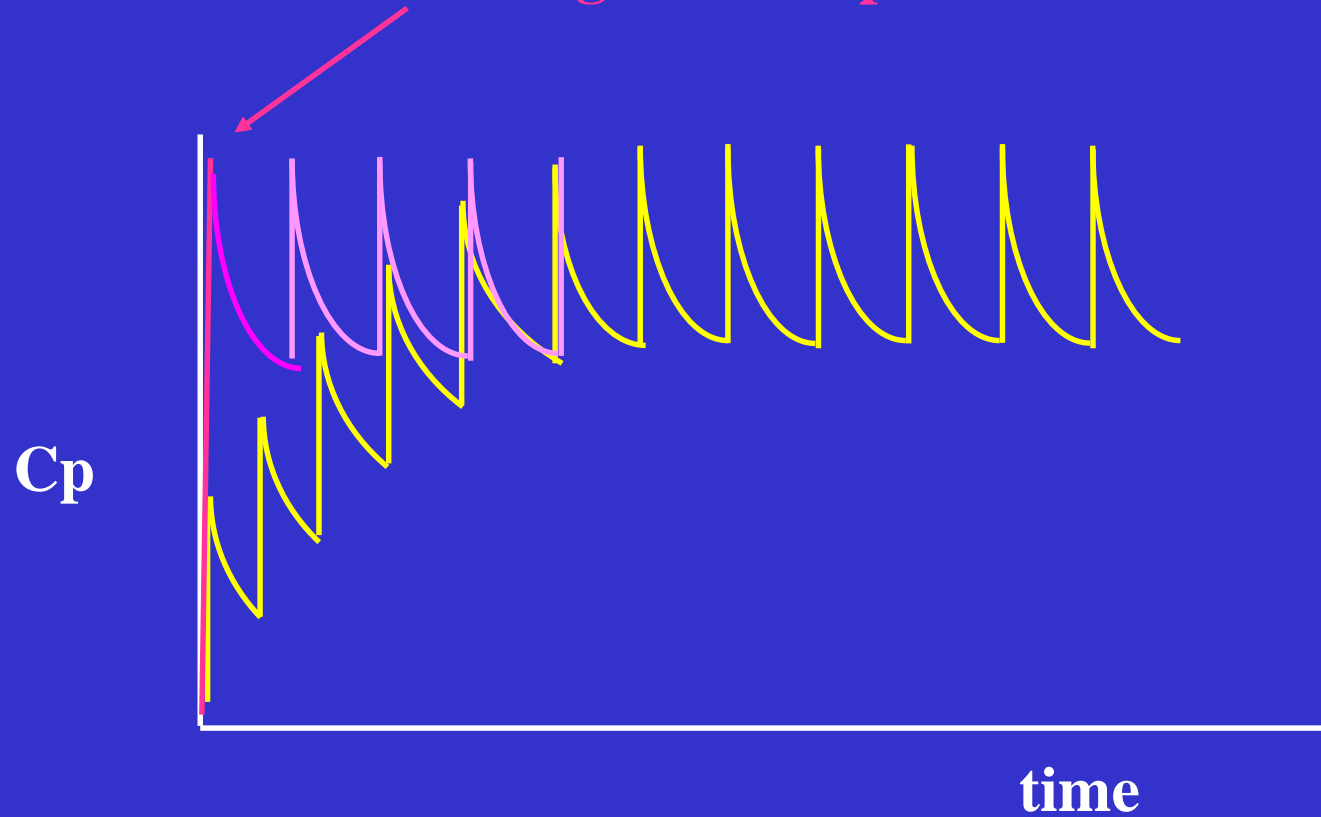
At Steady State

amount administered = amount eliminated between doses



Loading dose(s)

Loading dose = $C_{peak} \cdot \text{Volume of distribution}$



Therapeutic Drug Monitoring

Onset, Peak, and Duration of Action

Onset

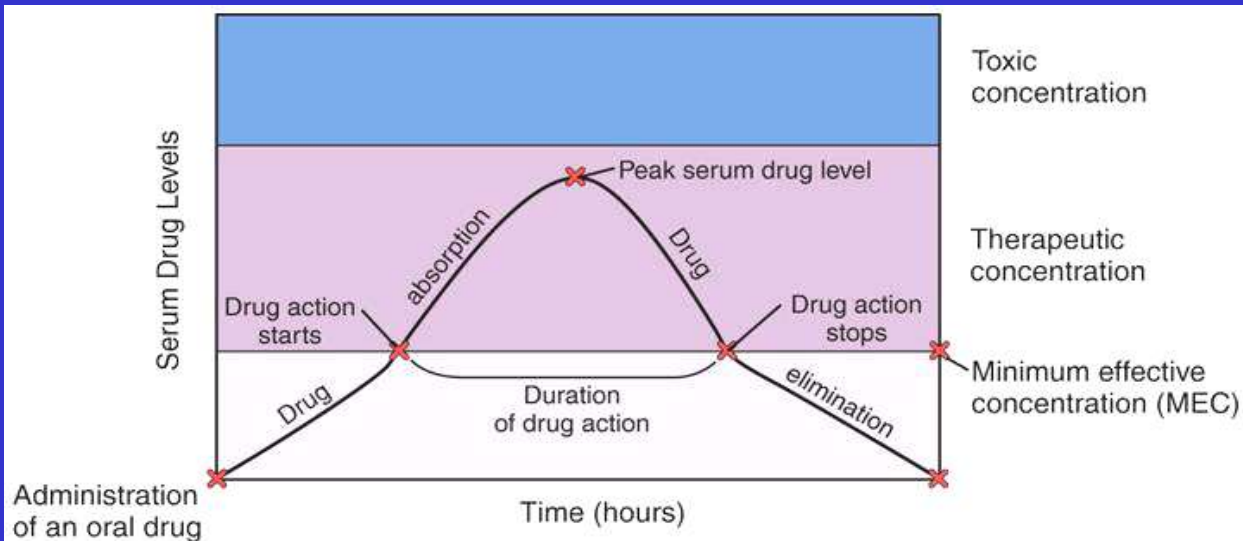
- The time it takes for the drug to elicit a therapeutic response

Peak

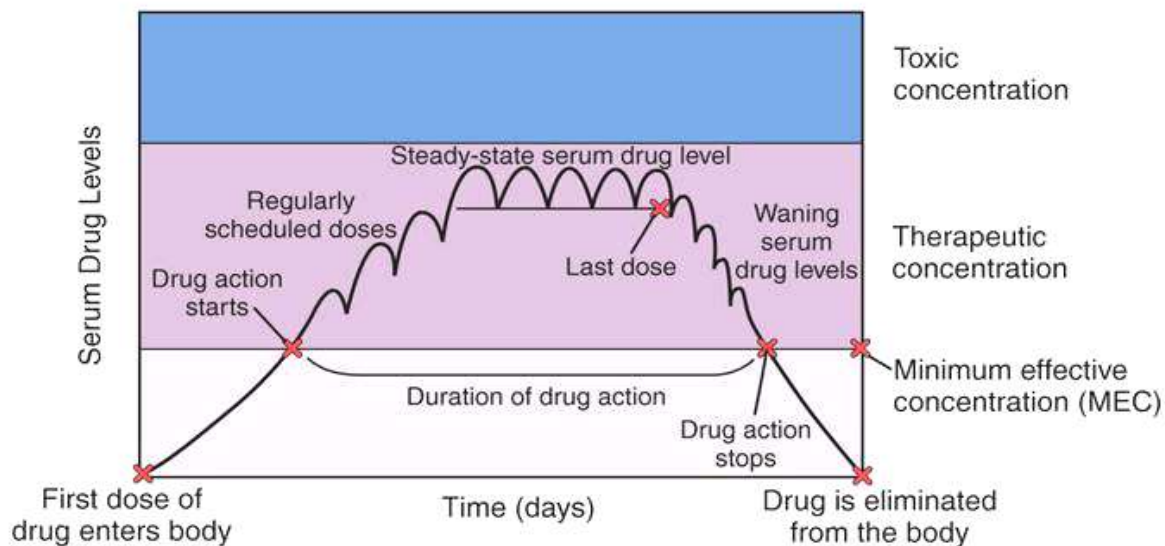
- The time it takes for a drug to reach its maximum therapeutic response

Duration

- The time a drug concentration is sufficient to elicit a therapeutic response



Drug action in relation to serum drug levels and time after a single dose.



Drug action in relation to serum drug levels with repeated doses.

Therapeutic Drug Monitoring

Peak level

- Highest blood level

Trough level

- Lowest blood level

Drug plasma concentration monitoring is helpful for drugs

- that have a low therapeutic index**
- that are not metabolized to active metabolites**
- whose concentration is not predictable from the dose**
- whose concentration relates well to either the therapeutic effect or the toxic effect, and preferably both**
- that are often taken in overdose**

For which specific drugs is drug concentration monitoring helpful?

The important drugs are:

- aminoglycoside antibiotics (plasma or serum)
- cyclosporine (whole blood)
- digoxin and digitoxin (plasma or serum)
- lithium (serum)
- phenytoin (plasma or serum)
- theophylline (plasma or serum)
- paracetamol and salicylate (overdose) (plasma or serum).

Other drugs are sometimes measured:

- anticonvulsants other than phenytoin (eg carbamazepine, valproate)
- tricyclic antidepressants (especially nortriptyline)
- anti-arrhythmic drugs (eg amiodarone).

The uses of monitoring are

- **to assess adherence to therapy**
- **to individualize therapy**
- **to diagnose toxicity**
- **to guide withdrawal of therapy**
- **to determine whether a patient is already taking a drug before starting therapy (eg theophylline in an unconscious patient with asthma)**
- **in research (eg to monitor for drug interactions in post-marketing surveillance using population pharmacokinetics).**

Altered pharmacokinetic profile

- **liver metabolism**

 - Disease**

 - Pharmacogenetics (cytochrome P450 polymorphisms)**

- **renal impairment**

 - Disease**

 - Elderly**